



bioenergy2020+

Characterisation methods for solid biomass fuels regarding their thermal utilisation in fixed-bed reactors

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A photograph showing laboratory glassware, including several Erlenmeyer flasks and beakers, some containing liquids, on a lab bench. The background is slightly blurred, showing what appears to be a laboratory setting.

Introduction

- **Main biomass groups: woody biomass, herbaceous biomass, fruit biomass (agricultural residues), and blends and mixtures**
 - Different physical behaviour and **chemical compositions** typical for **different biomass groups**
 - Higher amounts of ash, N and ash forming elements (e.g. S, Cl, Si, K, P) for herbaceous biomass compared to woody biomass
- **Increased utilisation of alternative biomass feedstocks (herbaceous biomass and agricultural residues) for combustion purposes can cause:**
 - Ash-related problems (ash melting on grates, deposit formation and corrosion)
 - Increased particulate, NO_x , SO_x and HCl emissions
- **One strategy to minimise problematic combustion behaviours is the application of fuel additives and fuel blends**
- **Up to now the evaluation of mixing ratios of fuel additives and fuel blends is mainly based on experiences and trial and error methods**



Objectives

- Development of characterisation methods in order to minimise the necessity for time-consuming and expensive real-scale combustion tests
- Development of characterisation methods/tools for solid biomass to predict combustion-related properties based on **laboratory approaches** and appropriate calculation procedures; focusing on fixed-bed combustion
 - Targeted selection of biomass mixtures and biomass additives to improve the combustion behaviour by **fuel indices, thermodynamic equilibrium calculations** (TEC) and **fixed-bed lab-scale reactor experiments**
- Development of a new **single particle reactor**
 - Determine relevant parameters during the thermal conversion process → simultaneously record **time resolved release profiles of relevant inorganic aerosol forming elements** (S, Cl, K, Na, Zn and Pb)



Fuel indices – Methodology

- **Fuel and ash analysis**
 - Only **highly accurate analytical methods** applied
 - Fuel sample has to be **representative** for the fuel of interest → different parts (e.g. leaves, stalks, grains, fruit) of a plant may show strongly deviating chemical compositions
- **Plenty of coal indices available → not applicable for biomass fuels**
- **Fuel indices can be used for a qualitative pre-evaluation of combustion related problems**
- **Fuel indices are based on results of chemical analyses, theoretical considerations, experiences from lab,- pilot- and real-scale combustion tests**
- **Relevant fuel properties are correlated with**
 - Gaseous and **particulate** emissions
 - Release ratios of relevant inorganic aerosol forming elements
 - Ash melting temperature
 - Corrosion risks
- **Validated by results of grate furnace combustion tests in the power range between 140 kW_{th} and 110 MW_{th}**



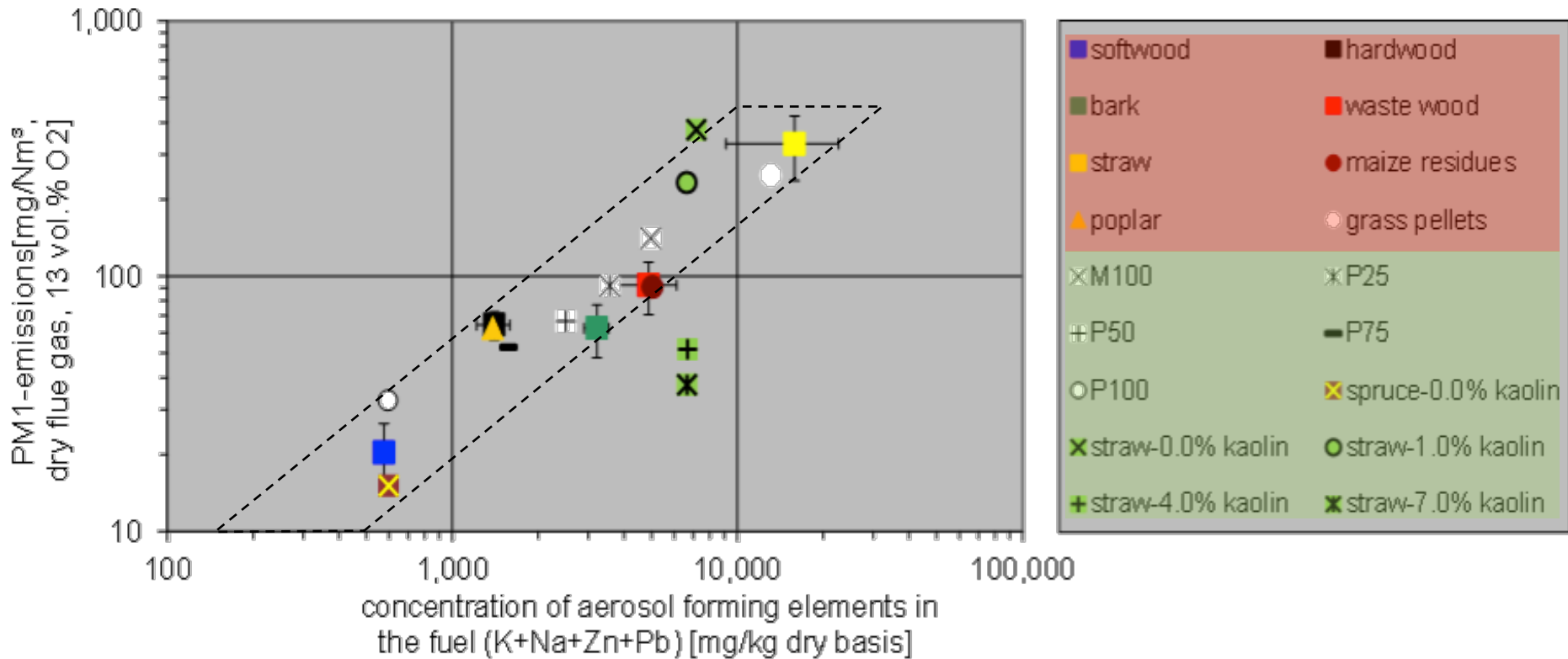
Fuel indices – Overview

■ Fuel indices that have been investigated/defined:

- N-concentration in the fuel as an indicator for NO_x emissions
- **Sum of K, Na, Zn and Pb** as an indicator regarding **aerosol emissions** and deposit build-up
- Molar Si/K ratio for an estimation of the K release from the fuel to the gas phase
- Molar 2S/Cl ratio as an indicator for high-temperature corrosion risks
- Molar $(\text{K}+\text{Na})/[\text{x}\cdot(2\text{S}+\text{Cl})]$ ratio to predict the gaseous SO_x and HCl emission range to be expected
- The prediction of the ash melting temperatures with the molar $\text{Si}/(\text{Ca}+\text{Mg})$, the molar $(\text{Si}+\text{K}+\text{P})/(\text{Ca}+\text{Mg})$ ratio as well as the molar $(\text{Si}+\text{K}+\text{P})/(\text{Ca}+\text{Mg}+\text{Al})$ ratio



Fuel indices – Example: Aerosol emissions



Explanations: PM1... particles smaller 1 μm (aerosols);

data origin: ■ full-scale combustion tests; ■ fixed-bed lab-scale reactor tests

- **With increasing concentration of K, Na, Zn and Pb in the fuel the aerosol emissions rises**
- **Index applicable for fuel blends, not valid for biomass/kaolin mixtures**

Fixed-bed lab-scale reactor – Description

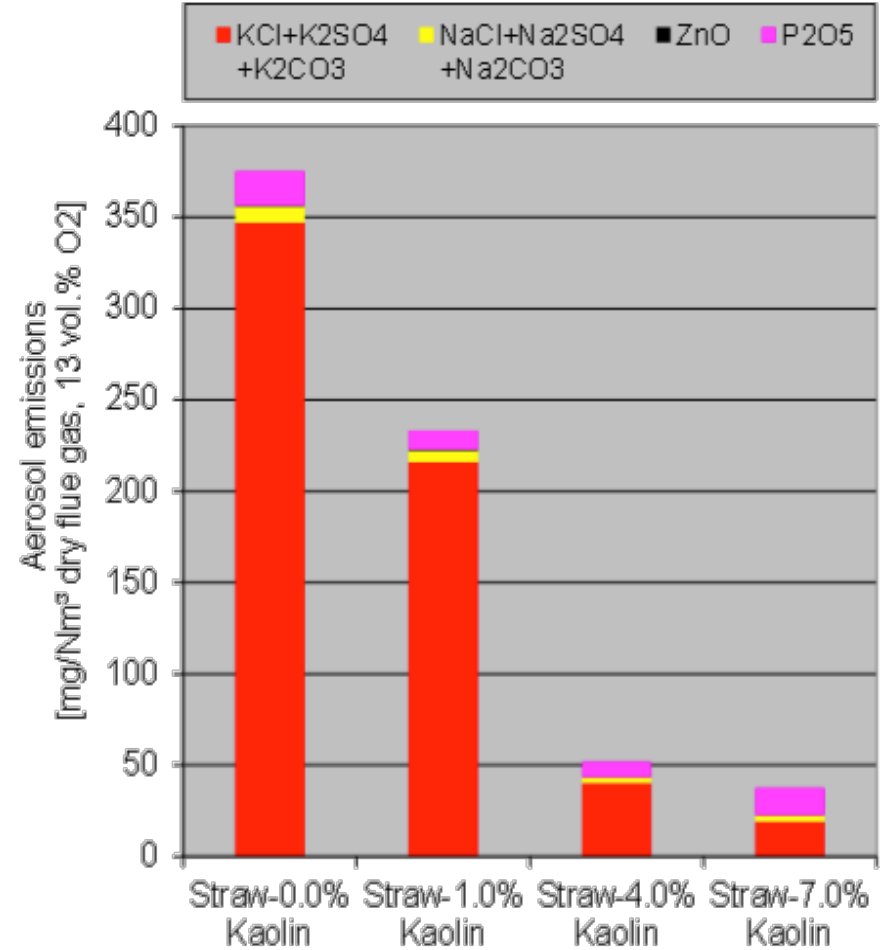
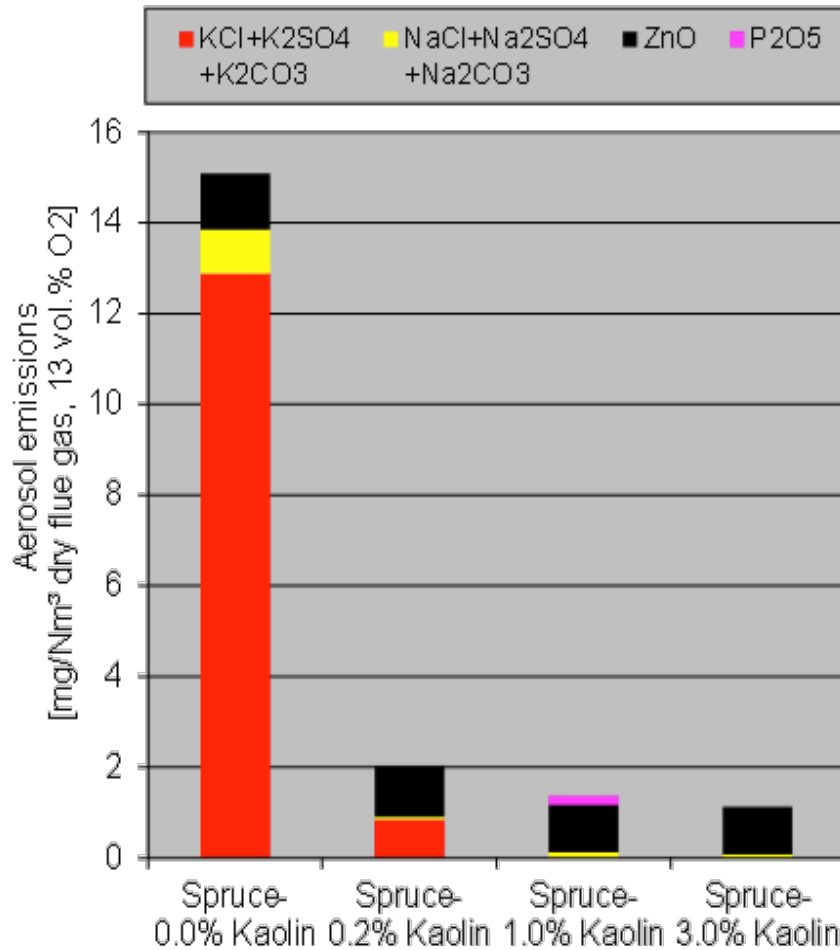


- Batch type reactor to **simulate** the fuel decomposition behaviour in real-scale grate combustion systems
- Tests with the lab-scale reactor provide results on:
 - Combustion behaviour
 - Release of NO_x precursors
 - **Release of aerosol forming elements** from the fuel to the gas phase
 - First indications about the ash melting tendency (optical evaluation)





Fixed-bed lab-scale reactor – Example: Prediction of aerosol emissions



Predicted aerosol emissions (by fixed-bed lab-scale reactor experiments) for pure spruce and spruce/kaolin mixtures as well as for straw and straw/kaolin mixtures

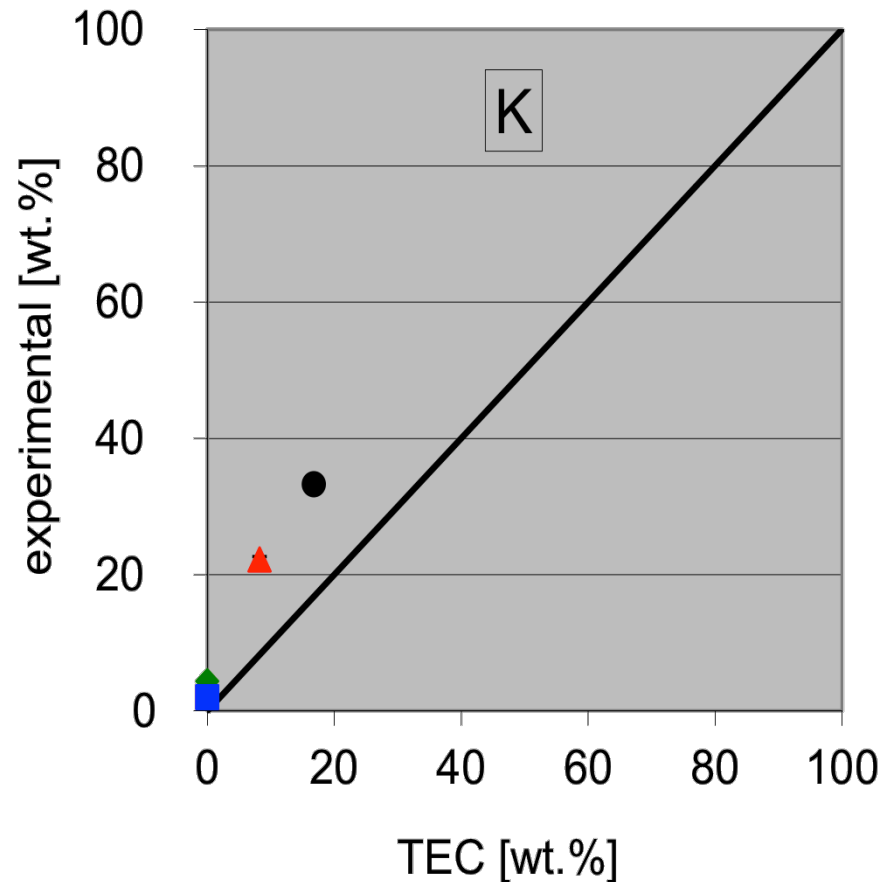
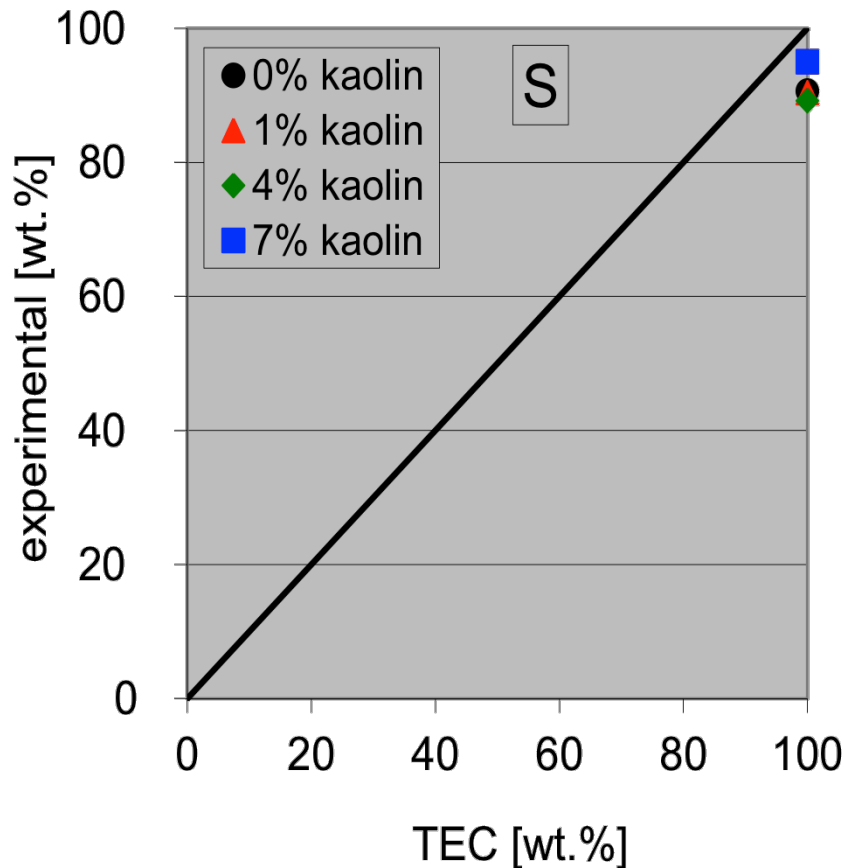
A photograph of laboratory glassware, including several Erlenmeyer flasks and beakers, some containing liquids, on a lab bench. The background is slightly blurred, showing a typical laboratory setting.

Thermodynamic equilibrium calculations – Description

- **Thermodynamic equilibrium calculations (TEC) provide information regarding:**
 - Ash composition and fractionation of individual elements in solid, liquid and gaseous phases
 - Composition of phases formed
 - **Release of relevant aerosol forming elements (S, Cl, K, Na, Zn, Pb)**
 - Ash melting behaviour



Thermodynamic equilibrium calculations – Example: Release of S and K



Experimentally determined element release of S and K by fixed-bed lab-scale reactor experiments in comparison with TEC for pure straw and straw/kaolin mixtures

Explanation: TEC .. thermodynamic equilibrium calculations

A photograph showing laboratory glassware, including several beakers and flasks, some containing white powders, on a lab bench. The background is slightly blurred, showing a typical laboratory setting.

Single particle reactor – Background & objectives

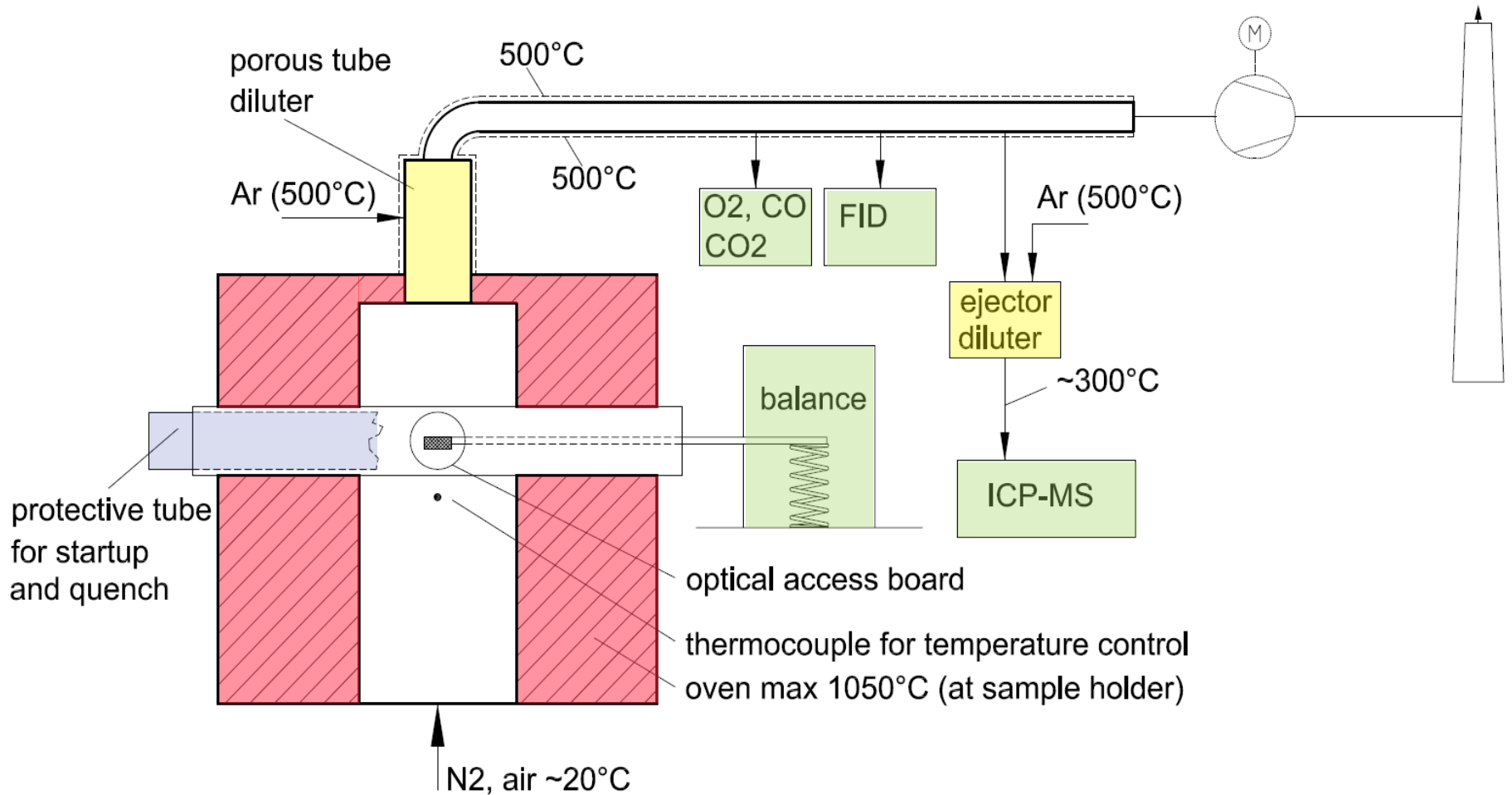
- **Release of inorganic components during biomass combustion/ gasification → special relevance:**
 - Aerosol formation, deposit build-up and boiler tube corrosion
- **Most studies describe the release during batch or real-scale experiments**
 - Integral release rate (entire combustion experiment)
- **Objectives of the single particle reactor development**
 - Development of a **single particle reactor** that enables a **characterisation** of the **combustion process** as well as an **online determination** of **volatile ash forming elements** released, by connecting the reactor to an inductively coupled plasma mass spectrometer (ICP-MS)

A photograph showing laboratory glassware, including several beakers and flasks on a stand, with a blue flame visible in the background, likely from a Bunsen burner.

Single particle reactor – Methodology

- **Reactor concept is based on a macro thermogravimetric analyser (TGA)**
 - Online recording of weight loss, temperatures
- **Batch type single particle reactor (SPR)**
- **2-step dilution of the sample gas**
 - **Porous tube diluter** → needed regarding the detection limits of the gas analyser
 - **Ejector diluter** → diluted gas flow to the ICP-MS
- **Online measurement of relevant aerosol forming elements:**
 - ICP-MS: Agilent Technologies series 7700x
 - Recorded intensities: ^{13}C , ^{34}S , ^{23}Na , ^{35}Cl , ^{39}K , ^{66}Zn , ^{132}Xe , ^{208}Pb
- **Quantification of the recorded intensities over the integral release rate**
 - Element balance for fuel and ash residues

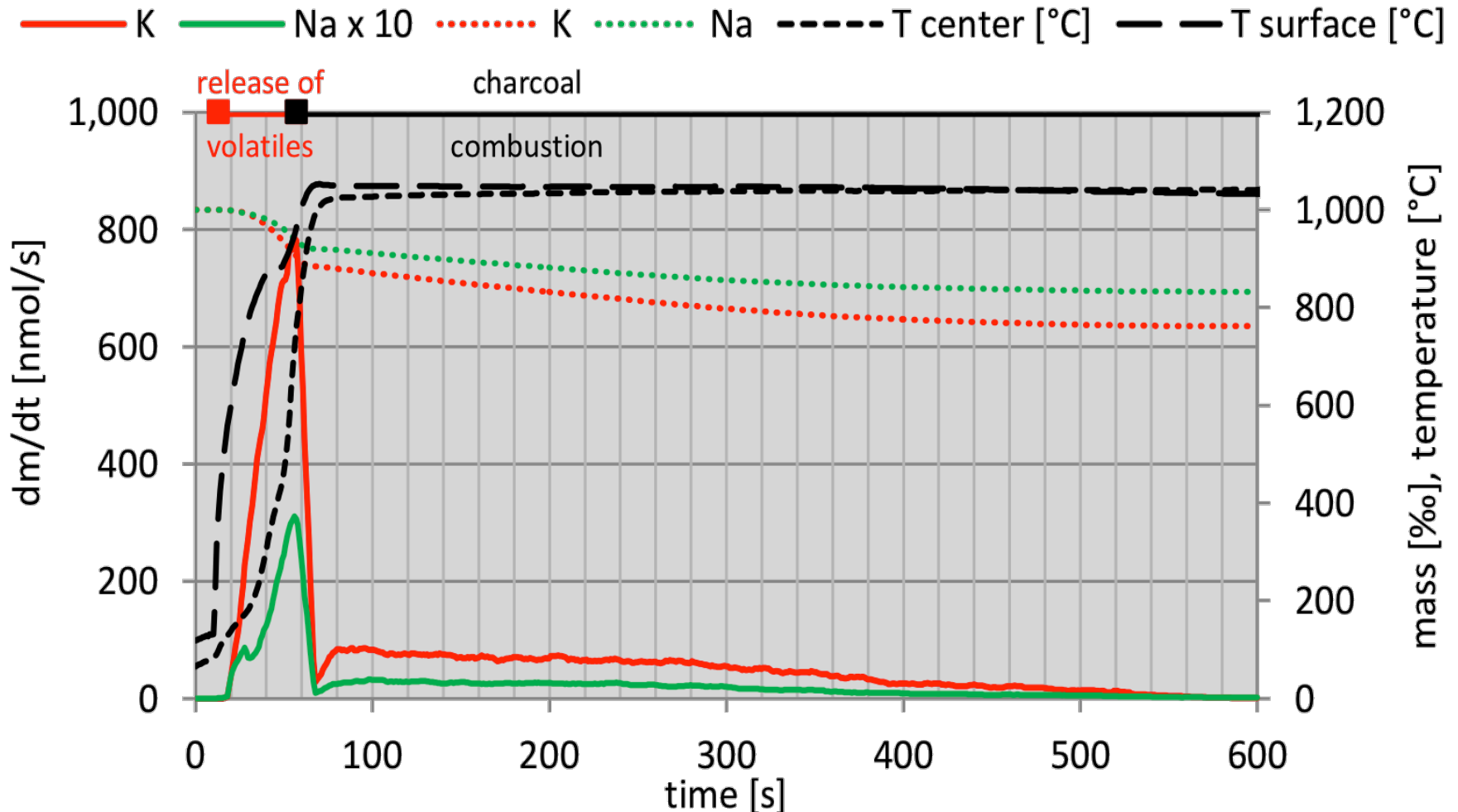
Single particle reactor – Complete setting



Explanation: FID ... flame ionisation detector, ICP-MS ... inductively coupled plasma spectrometer



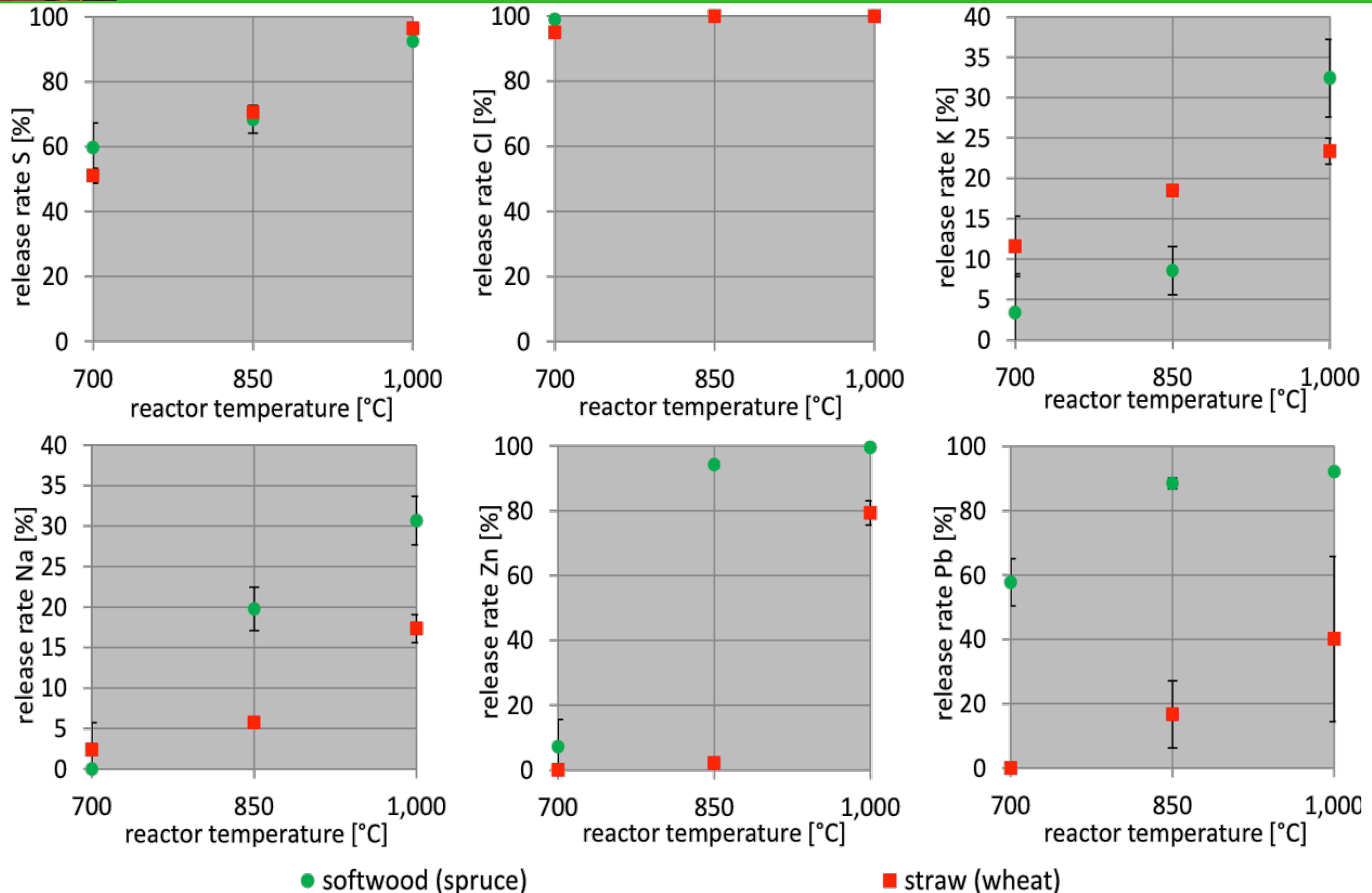
Single particle reactor – Example: Time resolved release behaviour of K and Na



Time dependent release of K and Na from straw at reactor temperatures of 1,000°C



Single particle reactor – Example: Release rates for softwood and straw at 6 vol.% O₂



Explanation: Release rates related to the element content in the biomass fuels

A photograph showing laboratory glassware, including several beakers and flasks, some containing white powders, on a lab bench. The background is slightly blurred, showing what appears to be a laboratory setting.

Summary and conclusions

- **Fuel indices** can be used for a **pre-evaluation** regarding the combustion behaviour
- **TEC** can be applied for a **qualitative determination** of the release of aerosol forming elements and the ash melting behaviour
- For a quantitative evaluation of the release of aerosol forming elements **fixed-bed lab-scale reactor** tests can be performed
- For inorganic element release in dependence of time and particle-temperature a new **single particle reactor** is available
 - More detailed insight into release behaviour and release processes possible
- This is a general approach for a **comprehensive fuel characterisation** also applicable for biomass additive mixtures and fuel blends
 - By these methods the need for **cost and time intensive full-scale combustion tests** can be **minimised**



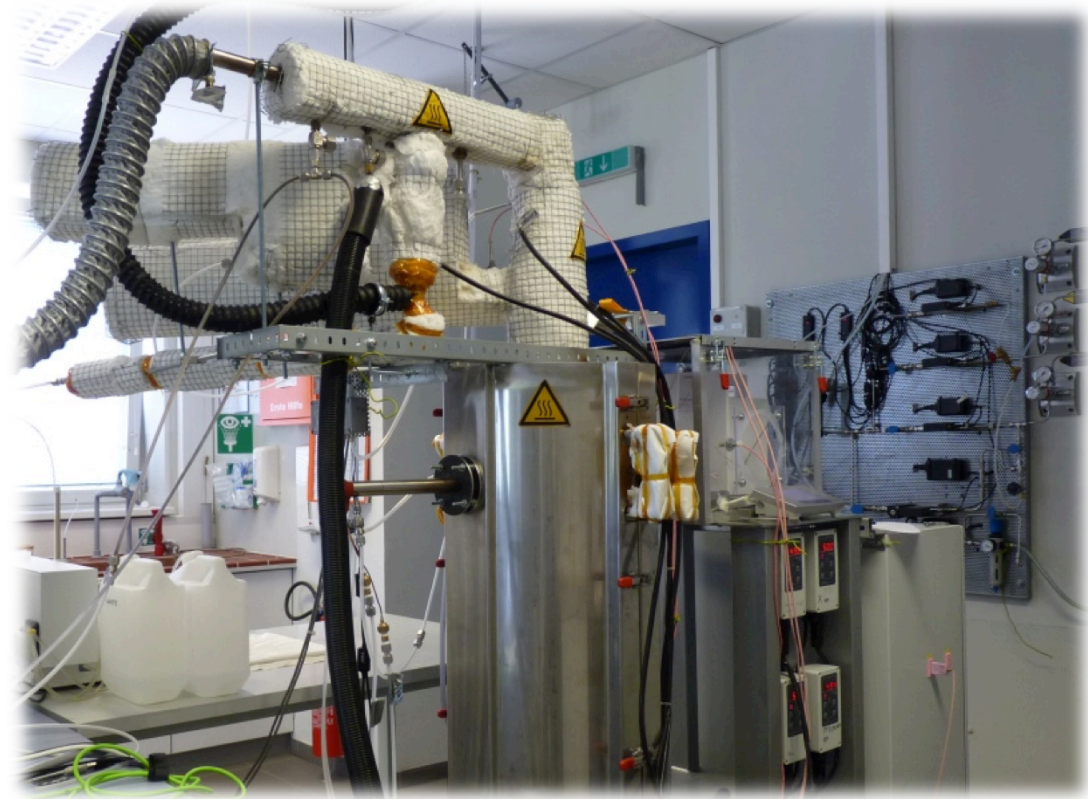
Thank you for your attention!

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